

Molecular dynamics and its applications  
CHEM 440 and 740/74100  
Winter 2012

Class Time: Tuesday and Thursday, 13:00–14:20

**Instructor** Pierre–Nicholas Roy  
**Office** ESC 330A  
**Office Hours** Tuesday and Thursday from 11:00 to 12:00 AM  
(or by appointment)  
**Phone** (519)–888–4567 ext. 38640  
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**Useful references:**

1. Allen and Tildesley, *Computer Simulation of Liquids*, Oxford, 1987.
2. Chandler, *Introduction to Modern Statistical Mechanics*, Oxford, 1987.
3. McQuarrie, *Statistical Mechanics*, Harper and Row, 1976.
4. K. Hinsen, *The molecular modeling toolkit: A new approach to molecular simulations*, J. Comp. Chem. 21(2), 79–85 (2000).

**Grade Distribution:**

Problem Sets (approximately 5)	15%
Midterm Exam (February 28; in class)	25%
Report and oral presentation (based on the critical assessment of a literature article)	10%
Final Exam	50%
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Total	100%

**Examinations:**

Excused absences from examinations will be given only if a completed Verification of Illness Form is provided. The form can be obtained from the Health Services web–site.<sup>1,2</sup> The completed form must first be registered with the Science Undergraduate Office (ESC 253). Absence for any other reason will result in a grade of ZERO. In the case of an excused absence on a midterm examination, the credit for the midterm will be transferred to the final examination.

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<sup>1</sup> <http://www.healthservices.uwaterloo.ca/>

<sup>2</sup> See equivalent service at Guelph

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**Proposed Course Outline**

1. Introduction: The purpose of molecular simulations
2. Statistical Mechanics primer
  - a. Ensembles, distributions, and partition functions
  - b. Properties: Ensemble averages
3. Interactions and force fields
4. Equilibrium
  - a. The Ergodic hypothesis
  - b. Molecular dynamics simulations
    - i. Classical Equations of motion
    - ii. Constraints
    - iii. Constant Energy simulations
    - iv. Constant temperature simulations
  - c. Monte Carlo simulations
    - i. Importance sampling (Metropolis method)
  - d. Structural properties and distribution functions
    - i. The structure of liquids
    - ii. Solvation
  - e. Quantum effects
    - i. Path Integral methods
    - ii. Application: equilibrium aspects of proton transfer
5. Dynamics
  - a. Onsager's regression hypothesis and time correlation functions
    - i. Green-Kubo relations and connection to transport properties
    - ii. Diffusion in liquids
    - iii. Reaction rate constants
  - b. Quantum effects
    - i. Wave-packet propagation
    - ii. Calculation of spectra